Methods for Detection and Quantification of Aflatoxins

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1. Introduction

Mycotoxins are fungal toxic metabolites which naturally contaminate food and feed. Aflatoxins (AFs), a kind of mycotoxins, are the main toxic secondary metabolites of some Aspergillus moulds such as Aspergillus flavus, Aspergillus parasiticus and the rare Aspergillus nomius (Ali et al., 2005, Alcaide-Molina et al., 2009). Such toxins can be separated into aflatoxins B1, B2, G1, B2a and G2a. Its order of toxicity is B1 > G1 > B2 > G2. Letters ‘B’ and ‘G’ refer to its blue and green fluorescence colors produced by these compounds under UV light. Numbers 1 and 2 indicate major and minor compounds, respectively (Weidenbörner, 2001; Hussein & Brasel, 2001). A. flavus only produces B aflatoxins, while A. parasiticus and A. nomius also produce G aflatoxins (Alcaide-Molina et al., 2009).

Aflatoxins are produced on various grains and nuts, e.g., corn, sorghum, cottonseed, peanuts, pistachio nuts, copra, cereals, fruits, oilseeds, dried fruits, cocoa, spices and beer in the field and during storage. AFs occur mainly in hot and humid regions where high temperature and humidity are optimal for moulds growth and toxins production (Ventura et al., 2004; Zollner & Mayer-Helm, 2006). Its presence is enhanced by factors as stress or damage to the crop due to drought before harvest, insect activity, soil type and inadequate storage conditions (Alcaide-Molina et al., 2009).

Aflatoxins, when ingested, inhaled or adsorbed through the skin, have carcinogenic, hepatotoxic, teratogenic and mutagenic effects in human and animals (rats, ferrets, ducks, trout, dogs, turkeys, cattle and pigs) (Anwar-Ul_Haq & Iqbal, 2004) even at very small concentrations. When aflatoxins B1 is ingested by cows, it is transformed into its hydroxylated product, AFs M1 and M2. Such aflatoxins is secreted in the milk and is relatively stable during milk pasteurization, storage, and preparation of various dairy products (Stroka & Anklam, 2002).

Among the more than 300 known mycotoxins, aflatoxins represent the main threat worldwide. After 1975 there has been an increased concern about the possibility of the presence of carcinogenic mold metabolites, particularly aflatoxins in food and animal feed products. Although aflatoxins are regulated in more than 80 countries, their legislation is not yet completely harmonized at the international level (Cucci et al., 2007). Several
institutions around the world have classified and regulated aflatoxins in food. The European Union (EU) has the most rigorous regulations concerning mycotoxins in food. The limits of AFB1 and total AF in foods are 5 and 10 µg/kg, respectively, in more than 75 countries around the world whilst they are 2 and 4 µg/kg in the European Union (EU) (Herzallah, 2009). The maximum residue levels for total AFs and also for the most toxic of them (AFB1) according to the EU Commission Regulations are 2 and 4 g/kg, respectively. The maximum legal limit for AFM1 in milk is set at 0.05µg/kg (50 ppt) for all EU Member States, and 25 ppt for baby food (Cucci et al., 2007). The European Committee Regulations (ECR) has established the maximum acceptable level of AFB1 in cereals, peanuts and dried fruits for direct human consumption in 4ng/g for total aflatoxins (AFB1, AFG1, AFB2, AFG2) and 2ng/g for AFB1 alone (Ricci et al., 2007). The International Agency for Research on Cancer (IARC) classified aflatoxins as Group 1 of human carcinogens (Alcaide-Molina et al., 2009). In USA, the U.S. Department of Agriculture and the U.S. Food and Drug Administration (FDA) have established an "actionable" level of 15-20 ppb of AFs in animal feed products.

Because of such facts, several methodologies for detection and quantification of AFs have been developed. The principal immunochemical based assay is the widespread enzyme linked immunosorbent assay (ELISA). Other methodologies base their performance upon electrochemical and optical principles such as: chromatography, UV-absorption, spectrometry, fluorescence and immunochemical assay tests. The aforementioned methods require well equipped laboratories, trained personnel, harmful solvents and several hours to complete an assay. Novel methods for detection of aflatoxins try to avoid these disadvantages. Among such novel methods, it can be found: biosensors, electrokinetics, electrochemical transduction, amperometric detection, and adsorptive stripping voltammetry. Each of the aforementioned methodologies has its own advantages and limitations according to sensitivity, easiness of use and cost-effectiveness. The objective of this chapter is to provide a general overview of the different methodologies to detect and quantify aflatoxins in the food analysis field.

2. Electrochemicals techniques

Aflatoxins can be measured by the use of electricity and electrochemical immunosensors. These immunosensors consist of a pair of electrodes (measuring and reference), implemented by using the screen-printing technique. The measuring electrode is coated with specific antibodies which will retain interest aflatoxins in the sample, whereas the other electrode (reference) is commonly made of a combination of Ag / AgCl.

The measurement procedure is similar to that carried out by the ELISA test (Enzyme Linked immunoabsorbent Assay). ELISA process is done by taking a sample of the substance to be measured and mixed with a known portion of conjugated aflatoxins with a special enzyme in a microtiter plate hole, and then it is inserted the measuring electrode. In this way, free aflatoxins in the sample compete for fill the places available (antibodies) in the measuring electrode. After some stabilization time, the measuring electrode is removed from the sample, washed with a buffer solution that removes all traces of the sample and leaves intact the electrode coating with aflatoxins that were captured but are not conjugate. After cleaning procedure, the electrode is introduced in a substrate solution that reacts with enzymes in aflatoxins conjugate, changing the electrical conductivity of the substrate depending on the amount of labeled aflatoxins antibodies attached to the electrode. Thus,
the greater will see the effect of aflatoxins marked, the lower the concentration of free aflatoxins in the sample.

However, the electrodes developed by Tan et al. (2009), were coated with conjugate aflatoxins instead of being coated with specific antibody, whereas the sample was mixed with the antibody. In this manner, some antibodies will be captured by free aflatoxins in the sample and some others by those attached to the electrode. Following that, the electrode is washed and it is placed into a solution with antibodies conjugated with alkaline phosphatase enzyme that binds to the antibodies that are bound to conjugate aflatoxins onto the electrode. After that, the electrode is immersed in the substrate solution in order that antibodies conjugate react and cause a change in electrical conductivity.

Some methods have been reported the use of simple electrodes (Rameil et al., 2010; Tan et al., 2009), while others have made use of multiple electrodes (Neagu et al., 2009; Piermarini et al., 2007), where the latter has shown to have advantages over the first in that: it is more user friendly; it is possible to carry out many experiments in parallel with different samples; and it reduces the time required for new procedures (Piermarini et al., 2007).

In order to measure the electrical conductivity in the electrodes there are different techniques, such as intermittent pulse amperometric (IPA), potentiometry, or linear sweep voltage (LSV).

The intermittent pulse amperometric technique involves the application of a periodic pulse of some duration fixed voltage across the electrodes coated and reference measurement, while the measured current varies depending on the conductivity of the substance. Moreover, in the technique of potentiometry, the measuring electrode coated is immersed in substrate solution without contaminating aflatoxins until a stable electrical potential is obtained, called the potential base. This potential varies depending on the amount of aflatoxins contained in the sample. In the linear sweep voltage technique, the sample is fed with a voltage which changes linearly, with a fixed slope.

The ability of these techniques to detect aflatoxins depends on many factors, including the type of substrate solution that is used, as is the case reported by Rameil et al. (2010), where it was shown that the use of 3 - (4-hydroxyphenyl ) propionic acid (p-HPPA), being a little toxic substance and does not require the use of organic solvents, can increase the conductivity of the substrate in potentiometry to measure aflatoxins M1 in milk. Another factor is the concentration of antibodies in the lining of the electrode, since the higher concentration of these, it can be reached higher peak current than in IPA technique, although the relationship between antibody concentration and electric current conducted is linear in a certain range, such as Tan et al. (2009) work suggests, where the linear range extends from a dilution of 1:30000 to 1:10000 of antibody against aflatoxins B1 found in rice, being the latter dilution which gave the best results.

Another point to consider is the detection limit, defined as the maximum decrease in signal equal to three times the standard deviation measured in the absence of aflatoxins to be determined. Detection limits down to 1 pg/ml have been obtained in the measurement of aflatoxins M1 in milk (Neagu et al., 2009); meanwhile, detection of aflatoxins B1 in rice has reached the limit 0.06 ng/ml (Tan et al., 2009).

There are also other measurement devices, as the case of piezoelectric immunosensors. Piezoelectricity is the property possessed by certain materials in which either generates a potential difference from applied mechanical deformation or vice versa (Webster 1999), so that materials that have this feature can resonate at certain frequencies. One of the most common piezoelectric materials is quartz crystal, used by Jin et al. (2009) as a sensor for
measuring aflatoxins B1 in milk. In this case, the crystal was treated to bind aflatoxins AFB1-BSA conjugate to the material for later subjecting to a similar procedure as mentioned by Tan et al. (2009), differing from this one in that the antibodies attached to conjugate aflatoxins attached to crystal, were marked with gold nanoparticles coated with antibody detector first. The concentration of aflatoxins will be reflected in this case as a change in the resonant frequency of the crystal, as reported by Jin et al. (2009) for the case of aflatoxins B1, where there is a linear relationship between the frequency of resonance and the logarithm of the concentration of aflatoxins.

3. Chromatography

Chromatography is one of the most popular methods to analyze mycotoxins such as aflatoxins. The most common techniques of chromatography are Gas chromatography (GC), liquid chromatography (LC), High performance liquid chromatography (HPLC) and Thin-layer chromatography (TLC). From these methods, LC and HPLC are the most used. In many cases, they are followed by fluorescence detections stage (Cavaliere et al., 2006). LC, TLC and HPLC are the most used quantitative methods in research and routine analysis of aflatoxins (Vosough et al., 2010); these techniques offer excellent sensitivities but they frequently require skilled operators, extensive sample pretreatment and expensive equipment (Sapsford et al., 2006).

3.1 Liquid chromatography

At the beginning the only separative method was GC, nevertheless, it is restricted to a small set of biological molecules for instance. Those should not be volatiles or should be derivatized (Roux et al., 2011). LC is other separative method which offers good sensitivity, high dynamic range, versatility and soft ionization conditions that permit access to the molecular mass of intact biological molecules. LC is usually coupled to fluorescence detection stage (FLD), UV absorption and amperometric detection (Elizalde-González, 1998) with pre-column derivatization or post-column derivatization stage (Cavaliere et al., 2006). LC coupled with fluorescence stage use the aflatoxins fluorescence properties to quantify them. So that, by improve this property it can be obtained better sensibility for aflatoxin detection. The most common techniques to improve fluorescence properties are the use of pre-column derivatization with trifluoretic acid and post-column derivatization with iodine or bromine (Elizalde-González, 1998). Other studies have been done in order to obtain enhancement of the fluorescence emissions of aflatoxins. Franco et al. (1998) collected emission data for AFQ1, AFM1, AFPI in solvents usually used for their chromatography separation in absence and in presence of different cyclodextrins. Such experiment was made in order to be applied principally in liquid chromatography.

3.2 Thin-Layer Chromatography (TLC)

Thin-layer chromatography is widely used in laboratories throughout the world for food analysis and quality control. Applications of TLC have been reported in areas of food composition, intentional additives, adulterants, contaminants, etc. TLC has been used to
analyze agricultural products and plants. It has advantages as: simplicity of operation; availability of many sensitive and selective reagents for detection and confirmation without interference of the mobile phase; ability to repeat detection and quantification; and cost effectiveness analysis, because many samples can be analyzed on a single plate with low solvent usage, and the time that TLC employs to analyze the sample is less than LC method (Sherma, 2000; Fuch et al., 2010). The most important differences between TLC and HPTLC are: the different particular size of stationary phase; the care used to apply the samples; and the way to process the obtained data (Fuch et al., 2010).

Diprossimo et al. (1996) present a work where show that TLC was superior to the methods of BF (Best food) CB-RCS-Mod (modified CB method-Rapid Modification of the Cottonseed Method) in terms of less fluorescence interferences, better solvent efficiency, and lower detection levels. Results obtained using TLC method compared to HPLC and enzyme-linked immunosorbent assay (ELISA) was found to agree among method but TLC was least expensive (Schaafsma et al., 1998).

Papers that use TLC methods to detect and quantify aflatoxins use sample clean-up based on immunoaffinity columns. Therefore, they avoid interfering compounds and allow visual quantification of aflatoxins at concentrations of less than 1 ng/g (Stroka et al., 2000). Immunoaffinity procedures provide very clean extracts because the sample is cleaned of interference substances. It also permits an easy aflatoxins determination, since they are applicable for automated sample clean-up (Stroka et al., 2002). Because of the advantages of this method, researches have been focused on them to develop new techniques to improve the methodologies for quantification of aflatoxins.

### 3.3 High Performance Liquid Chromatography (HPLC)

As aforementioned, HPLC is one of the most common methods to detect and quantify aflatoxins in food. It has been used jointly with techniques such as UV absorption, fluorescence, mass spectrometry and amperometric detectors. Elizalde-González et al. (1998) analyzed aflatoxins B1,B2, G1 and G2 based on HPLC and amperometric detection, and report that it is possible to detect 5 ng of all four aflatoxins. This proposed method is recommended for detection and quantification of the less toxic aflatoxin B2, which is presented in grains. Quinto et al. (2009) proposed a new method for determine aflatoxins B1, B2, G1, and G2 in cereal foods. This method is based on solid phase microextraction coupled with HPLC and a post-column photochemical derivatization to improve the fluorescence of analytes and fluorescence detection. Such method is fast compared with the complete analytical process that uses Immunoaffinity column. However, its sensitivity is below the legal limits. Vosugh et al. (2009) present a work that uses HPLC in conjunction with diode array detector (DAD) and a second order iterative algorithm called parallel factor analysis (PARAFAC). Such method is used for quantifying aflatoxins B1, B2, G1, and G2 in pistachio nuts, this work also use a solid phase extraction stage as a clean-up procedure. Manneta et al. (2005) presents a new method with fluorescence detection using pyridinium hydrobromide perbromide as a post-column derivatization agent to determine aflatoxin M1 in milk and cheese. The detection limits obtained were of 1 ng/kg for milk and 5 ng/kg for cheese that are 50-fold lower than the maximum residue level (MRL) for AFM1 in milk and 40-fold than MRL for AFM1 in cheese set by various European countries.

An interesting application of HPLC is the combination of immobilized enzyme reactor (IMER) in on-line high performance liquid chromatography. This combination allows the selectivity, rapidity and non-destructive, reproducibility of this chromatographic system to
be combined with the specification and sensitivity for an enzymatic reaction (Girelli & Mattei, 2005). Derivatization with a fluorophore enhances the natural fluorescence of aflatoxins and improves detectability. The pre-column approach uses the formation of the corresponding hemiacetals using trifluoroacetic acid (TFA), while the post-column one utilizes either bromination by an electrochemical cell or in addition of bromide, or pyridinium hydrobromide perbromide, for the mobile phase and the formation of an iodine derivative.

Even though the optical devices have dominated the traditional methods for HPLC, the present trend is to use mass detectors in the different HPLC types and configurations. This is because of the universal, selective and sensitive detection they provide (Alcaide-Molina, 2009).

There are several techniques that use chromatography for aflatoxin analysis in food (principally in milk, cheese, corn, peanuts, nuts). Commonly the quantification of the aflatoxins is made by a fluorescence detector that takes advantage of fluorescence properties of aflatoxins under determined wavelength. As a result, researchers have been focused on improving these fluorescence properties to develop more sensitive methods than the commonly used so far. Currently techniques such as pre-column derivatization and post-column derivatization are commonly used to improve aflatoxins fluorescence properties. They also have a clean-up stage to obtain a more pure sample, permitting a better quantification. Some of the common methods used in the clean-up stage are: immunoaffinity column and solid phase extraction.

### 3.4 Electrokinetics

HPLC is a method for detection of aflatoxins which often is enhanced by other techniques, resulting on alternative chromatographic methods. Accomplishing techniques related to electrokinetics are: Micellar electrokinetic chromatography (MEKC), reversed flow micellar electrokinetic chromatography (RFMEKC), and capillary electrokinetic chromatography (CEKC) with multiphoton excited fluorescence (MPE) detection, among others (Gilbert & Vargas, 2003).

Electrokinetics consists on an interfacial double layer of charges effect in heterogeneous fluids (Rathore and Guttman, 2003). Such effect generates the motion of the fluid due to an external force. This external force may be of different natures, but it is called electrophoresis when the force is an electric field; and capillary osmosis when the force is a chemical potential gradient and the motion of liquid happens in a porous body.

Capillary electrophoresis is a technique that although not been widely available as an alternative in many laboratories which routinely conduct HPLC, it has the advantage that it avoids the use of organic solvents. aflatoxin B1 can be determined by capillary electrophoresis (CEKC) with laser-induced fluorescence (LIF) detection (Maragos & Greer, 1997) after a clean-up process comparable to that required for HPLC, and with a very similar sensitivity to it. Besides, Electrophoresis does not require derivatization of aflatoxins, being that an advantage over HPLC. Sensitivity on CEKC can be further improved by using multiphoton excitation. Detection at levels 104 better than previously achieved by capillary separation in less than 90 seconds can be reached, which demonstrates the potential of this technique (Wei et al., 2000).

Micellar electrokinetic chromatography (MEKC) is conducted in polyacrylamide-coated capillaries under almost complete suppression of electroosmotic flow (Janini et al., 1996).
When small amounts of organic solvents are used in the buffer system good separation of aflatoxins are achieved. Nonetheless, it has been probed only with standard buffers (Gilbert & Vargas, 2003).

4. Fluorescence

All the aflatoxins have a maximum absorption around 360 nm (Akbas and Ozdemir, 2006). Letters ‘B’ and ‘G’ of the aflatoxins refer to its blue (425nm) and green-blue (450nm) fluorescence colours produced by these compounds under Ultra Violet (UV) light. AFB1 is the most common aflatoxin; it is followed by the AFB2. AFG is fairly rare. The fluorescence emission of the G toxin is more than 10 times greater than that for the B toxin (Alcaide-Molina et al., 2009).

Different techniques for detection of AFs related to fluorescence are exposed bellow.

4.1 Black light test

The black light test is a method which correctly identifies negative AFs samples with minimum expenditure of time and money. It consists on the illumination of the sample with a UV lamp. Tests should be made in a darkened area for best contrast. Fluorescence may be bright or dim, depending on the amount of fluorescing agent present. Polished metal surfaces reflect blue light, thus, users must be careful distinguishing fluorescence from such reflection. It is highly recommended to use safety goggles when working with the black light test. These goggles eliminate blue haze resulting from eye fluorescence caused by reflected longwave UV radiation.

However, fluorescence does not happen exclusively when aflatoxins are present. There are other substances in food that fluoresce under long wave UV radiation. Fungi as Aspergillus niger, various Penicillium species, Aspergillus repens and other species do not produce aflatoxins, but may produce fluorescent harmless metabolites. Then, it can be said that fluorescence is not a specific indication of the presence of aflatoxins, although it may indicate that conditions have been favourable for growth of toxic molds (B-100 Series Ultraviolet Lamps, UVP).

Furthermore, fluorescence is not stable. It disappears in 4 to 6 weeks of continuous exposure to visible or UV radiation although the toxin remains. Therefore, fresh samples must be taken. Hence, the reliability of the method depends on the size of the sample taken for analysis and how it is taken. A sample must be large enough to be representative of the entire lot and must be taken from all parts of the lot (B-100 Series Ultraviolet Lamps, UVP).

The black light test is commonly applied on animal feed. However, it is only a preliminary confirmatory test; it does not give a quantitative indication. Thus confirmatory and quantitative measurements are needed to be applied to those samples that reacted positively to the black light test. Non-fluorescing samples need not be subjected to this. A quantitative screening test which commonly follows the black light test is small chromatographic column (mini-column) (B-100 Series Ultraviolet Lamps, UVP). After the quantitative test a judgment can be made as to whether or not accept a lot.

4.2 Laser-Induced Fluorescence (LIF) screening method

LIF detection technique was pioneered by Yeung (Novotny & Ishii, 1985). This screening method consists on a mobile phase which contains an eluted sample of aflatoxins. Such
mobile phase passes through a detection window in the LIF detector. Thus, the whole fluorescence induced by the laser is collected by the detector (Alcaide-Molina et al., 2009). In LIF detection, the number of molecules that are photo-degraded is inversely proportional to the velocity of the fluorophore in front of the laser beam (Simeon et al., 2001). The scheme of a LIF sensor is shown on Fig. 1.

It has been said that AFB1 is the most toxic and one of the less fluorescent of the aflatoxins. However, the poorest sensitivity of the method may correspond to some other AF. Sensitivity tests should be applied for different AFs to select the one with the lowest sensitivity. The system should be calibrated with the curve of such aflatoxin; thereby, a signal provided by other AF is going to be translated into a higher concentration of this AF, leading to a confirmatory analysis on the screening method. This strategy, then, eliminates false negatives (Alcaide-Molina et al., 2009).

Thus, LIF detection shows as an appropriate detection technique with applications on very low concentrations of sample with native fluorescence or that fluoresce after derivatization (Simeon et al., 2001). However, LIF detection is a technique restricted to a limited number of laboratories because the high cost of the lasers, and because most of the analyte molecules have to be labelled with dyes that match the laser wavelength. Moreover, when the labelling reactions are not well understood, they can lead to contradictory results (Lalljie & Sandra, 1995).

Fig. 1. Scheme of a LIF detector (adapted from Simeon et al., 2001)
### 4.3 High-Performance Liquid Chromatography (HPLC) with LIF

Fluorescence detection and electrochemical detection are the two sensitive detection means most commonly used for quantitative studies in HPLC. This happens because the sensitivity levels of those hybrid techniques are much better than the ones observed with conventional fluorescence. It has been demonstrated the usefulness of LIF for sensitive detection in HPLC and micro high-performance liquid chromatography (µHPLC) in sensing very low concentrations of substances that can be excited in the near-UV range (325 nm) after labelling at nanomolar concentrations (Folestad et al., 1985; Diebold et al., 1979). Thus, LIF-HPLC method has become very popular and an essential detection technique in capillary electrophoresis (CE). Its sensitivity has been increased by the use of photoactivation devices (Reif & Metzger, 1995). Its popularity is due to its capability to detect substances at lower ranges than the micromolar (Bayle et al., 2004). For more information about HPLC refer to section 3.

It has been said that in LIF detection, the number of molecules that are photo-degraded is inversely proportional to the velocity of the fluorophore in front of the laser beam. On the other side, the sensitivity of detection in HPLC depends on the inner diameter of the capillary connected to the output of the column. Therefore, at a constant flow-rate, the sensitivity depends on the velocity of the fluorophore in front of the laser beam of the LIF, and the solid angle of fluorescence collection by the optical arrangement (Simeon et al., 1999). As a result, the union of LIF and HPLC offers a good compromise between sensitivity and dead volumes (Simeon et al., 2001).

In flow injection experiments with LIF-HPLC systems, at a given diameter, the detector signal will increase when increasing flow-rates if photochemical degradation is a limiting factor (Simeon et al., 2001). Conversely, if the flow-rate is fixed, an increase in diameter is expected to lead to a quadratic increase in the detector volume, generating also a quadratic increase in the number of detectable molecules. Then can be said that if a larger volume is irradiated at a larger capillary diameter, the efficiency of fluorescence collection is less important than in the case of smaller capillaries (Simeon et al., 2001).

### 4.4 Photomultipliers (PTM)

Since Fluorescence systems have a wide sensitivity, they are a useful tool to measure AFM1 in milk, which legal limit is very low (about 50 ppt). These systems are suitable for preliminary screening at the earlier stages of the industrial process, and make it possible to discard contaminated milk stocks before their inclusion in the production chain (Cucci et al., 2007). PMTs are highly sensitive photomultipliers based flow through detection system suited for ultra low fluorescence, chemiluminescence or bioluminescence measurements (PMT-FL, FIAlab Instruments). Their photon counting photon counting sensor has a blue-green (280–630 nm) spectral response with a peak of quantum efficiency at 400 nm and ultra-low dark counts. The high sensitivity of these devices reaches parts per trillion, permitting measurements of extremely low fluorescence signals. These devices may work with an internal excitation lamp, a LED source or an SMA terminated fiber optic cable for use with an external lamp. They also count with removable emission and excitation filters, allowing placing the most suitable emission filter for selecting the spectral region of interest. The output of the PMTs is expressed in photo-counts, and corresponds to the entire signal integrated in the transmission spectral band of the emission filter. Therefore, the signal acquired from a sample can also include a background contribution due to the solvent. In
principle, the latter can contribute to the actual fluorescence of the substance under analysis with a spurious signal of intrinsic fluorescence or Raman, depending on the excitation wavelength (Cucci et al., 2007).

The use of cyclodextrin (CD) as fluorescence enhancer for aflatoxins detection is widely reported in the literature (Zhilong, G. & Zhujun, 1997; Dall’asta et al., 2003), nevertheless, an increased error bar affects measurements due to the CD scattering effects.

The signal-to-noise ratio of these fluorescence measurements strongly depends on the type of cuvette used for containing the liquid sample. The cell geometry and its constituting material give rise to different effects, such as multiple reflections and stray-light. Small sample volumes and darkened walls are mandatory to achieve a better signal-to-noise ratio. Plastic cuvettes without the use of an additional fluorescence enhancer are not useful for the implementation of an early-warning system. Conversely, quartz cuvettes perform very well (Cucci et al., 2007).

Then, PTMs are compact and easy-to-handle sensors for the rapid detection of low concentrations of AFM1 in liquid solutions without the need for pre-concentration of the sample. They can be used as quick “threshold indications” and as an “early warning system”, so as to rapidly single out risk/alarm situations (Cucci et al., 2007).

5. **Ultra violet absorption**

It has been said that all the aflatoxins have a maximum absorption around 360 nm with a molar absorptivity of about 20,000 cm$^2$/mol (Akbas & Ozdemir, 2006). But, even though aflatoxins could be detected by UV absorption, the sensitivity of such systems is not sufficient to detect these compounds at the parts per billion (ppb) levels required for food analyses (Alcaide-Molina et al., 2009). The detection limit of UV sensors reaches micromolar ranges (Couderc et al., 1998). This is why fluorescence (FL) techniques have become more popular for AFs detection.

For overcoming the named limitation, UV absorption technique is usually combined with HPLC systems. Experimental results indicate that the detection limit of aflatoxins is enhanced by the proper method of extraction and clean-up process (Göbel & Lusky, 2004; Ali et al., 2005). For example, the selected clean-up and extraction procedures should minimize the interfering substances and matrix effect on the elution and separation of aflatoxins (Akiyama et al., 2001). Such important factors, correctly applied, may be of great importance to help the less sophisticated laboratories with HPLC instruments equipped with UV detector to detect aflatoxins with a precision that complies with the international guidelines and regulations.

Then, even though, HPLC-UV systems still are less sensitive than HPLC-FL systems, especially at low AF levels (Herzallah, 2009), HPLC-UV systems indicate to be accurate, precise, and consequently, reliable enough for determination of aflatoxins in food, with low duration and running cost.

6. **Spectrometry**

6.1 **Ion mobility spectrometry**

The Ion mobility spectrometry is a technique that is used in the characterization of chemicals on the basis of speed acquired by the gas-phase ions in an electric field. This technique has been used to determine the concentration of aflatoxins, as evidenced by the work of Sheibani
et al. (2008) in which are detected and quantified the concentration of aflatoxins B1 and B2 in pistachio. It has certain advantages in common with the FT-NIR, and low detection limit, fast response, simplicity, portability, low cost.

To detect aflatoxins in a sample, this is evaporated and mixed with a carrier gas. Then it is entered into the Ion Mobility Spectrometer (IMS) where the mixture is ionized and passed through an electric field gradient, where ions of different substances will travel at different speeds. The study by Sheibani et al. (2008) shows that using this technique is impossible to quantify as low as 0.25 ng.

6.2 Fourier Transform Near Infrared (FT-NIR) spectrometry

This technique has been underutilized for the detection of aflatoxins due to calibration requirements required against standard reference chemical processes (Tripathi & Mishra, 2009). Despite of the aforementioned limitations, this technique has some advantages, such as: fast and easy equipment operation, good accuracy, precision, performing nondestructive analyzes, prediction of chemical and physical sample from a single spectrum parameters from a single spectrum enabling several components to be determined simultaneously based on the use of multivariate calibrations.

It basically consists of measuring the absorbance of the sample to light whose wavelength varies in the range known as the Near Infrared (NIR). In the work of Tripathi & Mishra (2009) it is found that for the correct quantification of aflatoxins B1 in chili powder network readings were taken in the range of 6900.3 - 4998.8 cm\(^{-1}\) and also in the range of 4902.3 - 3999.8 cm\(^{-1}\), excluding the water absorption bands (5155 and 7000 cm\(^{-1}\)). Good results were obtained with respect to chemical techniques such as High Performance Liquid Chromatography (HPLC) and Thin Layer Chromatography (TLC), although its detection range is between 15 to 500 mg / kg which is slightly high compared to these techniques.

7. Biosensors

The term biosensors refers generally to a small, portable and analytical device based on the combination of recognition biomolecules with an appropriate transducer, and able of detecting chemical or biological materials selectively and with a high sensitivity (Paddle, 1996). Its principle of detection is the specific binding of the analyte of interest to the complementary biorecognition element immobilized on a suitable support medium. When the analyte binds the element, there happens a specific interaction which results in a change of one or more physico-chemical properties. Such properties may be: pH, electron transfer, mass, or heat transfer that are detected and can be measured by a transducer. Depending of the method of signal transduction, biosensors can be divided into different groups: electrochemical, optical, thermometric, piezo-electric or magnetic. In the case of aflatoxin detection, electrochemical and optical are the most commonly used (Velasco-Garcia & Mottram, 2002). Until 1996 only few biosensors for toxins were recorded and most of them were based on ELISA. The goal of the more recent studies is to simplify and expedite the method of detection while maintenance and improvement of sensitivity is attempted (Sapsford et al., 2006).

A method that has gained popularity is the use of antibodies to clean-up samples prior to measurement by LC of HPLC. Carlson et al. (2000) present an immune-affinity fluorometric
biosensor where the sample was filtered through a column containing sepharosa beads to which the polyclonal aflatoxin-specific antibodies were joined. The beads with attached aflatoxins were subsequently rinsed to remove any impurities and interference. Posterior, an eluant solution was passed through the beads causing antibodies to release the bound aflatoxins. The analyte was collected and placed in a fluorometer. This system consists essentially of two subsystems a fluidics subsystem in charge of mechanical-handling and processing and an electro-optical system that add a miniature fluorometer.

Sapsfor et al. (2006) present a system to detect and quantify foodborne contaminants using an array biosensor. It is capable of measuring large pathogens such as the bacteria Campylobacter jejuni and small toxins (mycotoxins ochratoxin A, fumonisin B, aflatoxin B₁ and deoxynivalenol). The system is capable of multiple detections of aflatoxins in a short time.

Aflatoxins have inhibitory effect on acetylcholinesterase (AchE) and their detection is coupled with the decrease in the activity of AchE which is measured using a choline oxidase amperometric biosensor (Nayak et al., 2009). Amperometric methods allow the detection of low aflatoxin concentration that cannot be detected by classical spectrophotometry because of the omission of the dilution step used in classical method.

Wang et al. (2009) present an implementation of Long range surface Plasmon – enhanced fluorescence spectroscopy (SPFS) in an immunoassay based biosensor for the highly sensitive detection of AFM1 in milk samples (LRSP). Here fluoropore-labeled molecules captured on the sensor are exited with surface plasmons (SPs) and the emitted fluorescence light is measured. The system takes the advantage of the electromagnetic intensity improvement occurring upon the resonance excitation of SPs that increase the intensity of fluorescence signal. This technique is based on surface Plasmon resonance which is becoming popular for the detection of chemical and biological species.

Others tendencies are the use of nanotechnology to detect aflatoxins such as the paper presented by Xiulan et al. (2004) where colloidal gold particles and antibodies were combined and used to develop an immunochromatographic (IC) method for aflatoxin B1 analysis. The result of this was that the analysis could be completed in less than 10 minutes and the lower test limit was 2.5ng/ml for aflatoxin B1. Such limit was increased in two times of ELISA.

When aflatoxins are consumed by cattle, they are transformed into their hydroxylated product, AFM1 that is known for its hepatotoxic and carcinogenic effects. To date, aflatoxins are regulated in many countries because of the milk intake in infants is high and when they are young the vulnerability to toxins is higher. Because of this, it is necessary to monitor AFM1 in milk at ultra low level, so that, analytical methods with high detectability and analytical throughput are required. Kanungo et al. (2011) present a novel approach where a highly sensitive microplate sandwich ELISA was developed and integrated with Magnetic nanoparticles (MNPs) which could detect ultra trace amount of AFM1 in milk. Sandwich-type immunoassay is an effective bioassay due to the high specificity and sensitivity. MNPs were used as an affinity capture column wherein immobilized antibodies on their surface could capture AFM1 from milk sample.

According to the aforementioned, the new trends could be the use of nanoparticles in combinations of the commonly used techniques such as LC, HPLC, TLC and immunoassay techniques. These combinations are to improve the detection at ultra low level of compounds in order to diminish the risk that this kind of mycotoxins causes to
humankind. For doing this it is necessary to use methods that combine simplicity with high detectability.

8. Adsorptive stripping voltammetry

Adsorptive Stripping Voltammetry is a method based on accumulation and reduction of AFB1 and AFB2 species on the surface of hanging mercury drop electrode (HMDE). Such electrode offers both sensitivity and selectivity. The pioneers on this method applied to detection of aflatoxins are Hajian and Ensafi (2009), for more information refer to their article.

Voltammetry is an electro-analytical method. It obtains information about the sample by measuring a current while the potential is varied (Komorsky et al., 1992). The voltammetry used in the experiment of Hajian and Ensafi had three-electrodes containing hanging mercury drop electrode as a working electrode, a carbon rod as an auxiliary electrode and an Ag/AgCl (3.0 M KCl) reference electrode. This method was proved only on AFs B1 and B2, where both aflatoxins were found to adsorb and undergo irreversible reduction reaction at the working mercury electrode (Rodriguez et al., 2005).

Adsorptive Stripping Voltammetry is an electrochemical method which has no or very low dependence on pH. This dependence displayed only for B1 in the pH range of 5.0 to 6.0 (Sun et al., 2005).

As it is expected in adsorption processes, by increasing accumulation time, the peak currents for both of the aflatoxins are increased and then leveled off because of the saturation of electrode surface (Hajian & Ensafi, 2009). Therefore, an accumulation time of 60 seconds is recommended for improving sensitivity. It is also recommended to use the extraction and clean-up method for aflatoxins that was used by Garden and Strachan (2001). Such extraction and clean-up method try to obtain the highest yield of aflatoxins with the minimum matrix effect.

This method uses single standard addition method by spiking 10 ng / ml of standard aflatoxin followed with general procedure for voltammetric analysis. The total determination of aflatoxins is based on the next formula reported by Hajian and Ensafi (2009):

$$Aflatoxin \left( \frac{ng}{ml} \right) = \frac{P'}{P} * C * 20$$

Where: $P'$ is peak current of sample (nA), $P$ is peak current of standard aflatoxin (after subtract from $P'$) (nA), $C$ is the concentration of aflatoxin spiked in the cell (ng/ ml) and 20 is a factor value after the sample weight, volume of methanol/water used in the extraction and preparation of injection sample have been considered (Hajian & Ensafi, 2009).

Adsorptive Stripping Voltammetry is a suitable method for determination of total aflatoxins (B1 and B2) in food. This method has some advantages such as high sensitivity, extended linear dynamic range, simplicity and speed (Hajian & Ensafi, 2009). The reliability of this method for determination of total aflatoxins is comparable to HPLC.

9. Miscellaneous methods

Different techniques have tried to offer new options for screening aflatoxins. Screening consists on rapid and/or in situ detection. There are two main difficulties for an effective
screening method: the necessity of a very high sensitivity, which in fact is a necessity of any technique; and the demand of preliminary sample preparations. Some of these techniques, which are commented ahead, present a lack of applications because of their practical inconveniences or because they have not been proved yet with real samples (Gilbert & Vargas, 2003).

**Optical-fiber:** Modular separation based on a fiber-optic sensor (Dickens & Sepaniak, 2000) has been tested in buffers, showing enough sensitivity (0.005 ng/ml for detection of aflatoxin B1). Unfortunately, it is limited to handling only liquid matrices.

**Electrochemical transduction:** The interaction of the aflatoxin M1 with bilayer lipid membranes can be sensed electrochemically (Andreou & Nikolelis, 1997; Andreou et al., 1997) reaching a good specificity and speed of response. But, its principal negative factor is its detection limit 750 ng/ml, which is very unpractical.

**Flow injection monitoring:** Stabilized systems of filter-supported membranes are capable of achieving significantly improved sensitivity (Andreou & Nikolelis, 1998). These membranes have been proposed for use in detecting aflatoxin M1 in cheese (Siontorou et al., 2000). Single strand DNA oligomers have been incorporated into the membranes to control surface electrostatic properties. This incorporation led to achievement a sensitivity much closer to regulatory limits, and with the ability to analyze four cheese samples per minute. Even though this technique appears to be a good option for in situ testing, it does not have yet many applications (Gilbert & Vargas, 2003).

### 9. Conclusion

Different methods for detection and quantification of aflatoxins have been discussed along this document. Through the researching made for this document, it has been found that the most popular methods are: ELISA, electrochemical immunosensors, chromatography and fluorescence. Even though ELISA is the most common and widespread technique, it has the disadvantage of requiring well equipped laboratories, trained personnel, harmful solvents and several hours to complete an assay. The detection and quantification of aflatoxins by using electrochemical immunosensor has proven to be efficient, easy to use and able to detect very low levels of these substances. Chromatography is a method which needs immunoaffinity columns and phase solid extraction need to be used to clean-up the sample, and also pre-column and post-column derivatization to enhance the aflatoxins fluorescence properties. So that, by improving these characteristics, it is possible to obtain a better quantification and sensibility. Fluorescence detection is a very good alternative to the conventional techniques used today. It has a very high sensitivity, especially when is combined with other techniques as HPLC. Some fluorescence techniques are used even in portable sensors, resulting on in situ measurements. Techniques such as FT-NIR spectrometer and IMS have proven to be quick, inexpensive and user-friendly, however, the FT-NIR technique shows lack of sensitivity when detecting low concentrations of aflatoxins. New techniques in this field are being developed in order to give a rapid and/or in situ detection of these toxins. Some examples of these new techniques are: optical-fiber, electrochemical transduction, low injection monitoring and biosensors. All of these, except for the biosensors, still present a lack of applications because of their practical inconveniences. The biosensors have been designed to overcome the drawbacks that the common tools employed to detect and quantify aflatoxins presents. They use the inherent fluorescence property that aflatoxins have to improve the detection, that in combination
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with optical and immunochemical techniques used to clean-up the samples achieve a better quantification. Due to the risk that the aflatoxins represent to humans, the researchers all over the world are looking for methods to detect and quantify them. Apparently, the measurement of aflatoxins in the future tends to be the combination of optical, immunochemical, and fluorescence techniques.

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11. References

B-100 Series Ultra Violet Lamps, Ultra Violet Products (UVP). Nuffield Road, Cambridge, UK.
PMT-FL fluorometer, FIAlab Instruments. Bellevue WA, USA.


This book is divided into three sections. The section called Aflatoxin Contamination discusses the importance that this subject has for a country like the case of China and mentions examples that illustrate the ubiquity of aflatoxins in various commodities. The section Measurement and Analysis describes the concept of measurement and analysis of aflatoxins from a historical perspective, the legal, and the state of the art in methodologies and techniques. Finally, the section entitled Approaches for Prevention and Control of Aflatoxins on Crops and on Different Foods, describes actions to prevent and mitigate the genotoxic effect of one of the most conspicuous aflatoxins, AFB1. In turn, it points out interventions to reduce identified aflatoxin-induced illness at agricultural, dietary and strategies that can control aflatoxin. Besides the preventive management, several approaches have been employed, including physical, chemical biological treatments and solvent extraction to detoxify AF in contaminated feeds and feedstuffs.

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