Chapter from the book *Water Quality Monitoring and Assessment*
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1. Introduction

Water has a fundamental impact on the economy and it is at the center of strategic interests. It is is an essential element of daily life for each and every one of us and a vital source of wealth and can become a possible destabilizing factor for the countries of the disinherit ed areas. The water provision is becoming more and more alarming which is due the galloping demography and the increasing droughts during these last years. In 2002, 1.5 million humans did not have access to safe drinking water and 2.5 million individuals in the world did not benefit from cleansing service (Phiri, 2005), (Lefort, 2006).

Algeria is one of the most underprivileged areas in the world with regard to hydrous availability (A. N. B. 2001). Furthermore, the demographic explosion and the economic growth led to a very high water demand which by far exceeds the available resources (Ajayi, 2002), the annual availability of water in east-Algeria is 600 m³/inhabitant year. This places it in the poor category countries in hydrous resources with respect to the rarity threshold level, fixed by the World Bank, of 1000 m³/inhabitant year (A. N. B. 2001), (Bouchentouf, 1994).

The strategy of any developing region in a country must necessarily be based primarily on "The control of water resources," they must be translated into a rational, continuous and sustainable water potential in the socio-economic order. Faced with ever-increasing challenges of water resource management in the country, a coherent politic of water at the national level is essential. Where with the development of the urbanization and the industrialization, as well as the evolution of consumption modes the discharges of waste water were considerably evolved in quantity and in quality. So sewer systems collect all the varieties of discharges such as domestic, industrial and commercial discharges with very diverse characteristics. The aim of the wastewater treatment was to reduce the pollution load they carry in order to protect the receiving natural environment such us sea water, oued...etc. and eventually for its future uses (fishing, leisure, food, agricultural or industrial use, etc.).
From this perspective our contribution will be particularly focused on the impact of residual treated wastewater from the sewage treatment plant (STEP) of Hennencha (Souk Ahras) on the water quality of Oued Medjerda, Principal River in the region, and eventually of Ain Dalia Dam.

Souk Ahras city is located at north east of Algeria and extends on a surface of 4360 km$^2$. It has four dams and an exclusive dependence on Medjerda River for its provision (ANRH, 2003). This natural water source is the principal river of the Maghreb region by its flow length as well as its area and the water volume procured (Derradji, 2004), (Louis, 1956).

2. Materials and methods

2.1 Study area

The low plain of Medjerda is situated in the north east of Algeria, at 680 km from Algiers and at 100 km from Annaba. The Medjerda River crosses the territories of two states: Algeria in its high part and Tunisia in its average and low parts. It takes its source from Khemissa, runs towards the east before discharging its contents in the Mediterranean sea, in the “golf of Tunis” (Figure. 1). It extends on 416 Km, 106 Km of which depend on the authority Souk Ahras city (Athmani, 2005). The river displays a flow that widely varies with the season, ranging from 1000 m$^3$/s at winter to 1 m$^3$/s at summer (A.B.H. 2001). The population of the studied area reaches more than 120000 inhabitants.

Fig. 1. Localization on map of Medjerda River (Elkenedj, 2009).
Realized in 1991 for a capacity of 200 m$^3$/day, the treatment plant of waste water (STEP of Henancha) receives currently more than 774 m$^3$/day causing serious water quality problems in Oued Medjerda which is the first receiver of the STEP rejections and also on the water quality of Ain Dalia dam located at 600 m downstream from STEP.

Geographical location:

The commune of Hennencha is located at 15 km at the West of Souk Ahras city. It is localised with the Lambert coordinates (Northern Algeria): X=96,977Km; Y=340,349Km figureNo 2.

2.2 Sampling

Samples from the Medjerda River were taken from sites A and E respectively located upstream and downstream the STEP. However, both sites are situated upstream the Ain Dalia dam. The sampling procedure was carried out once a month from Mars to July 2009.

2.3 Sampling procedure and material

The samples were taken in PVC 2-liter bottles. Before use, the bottles were treated with nitric acid, rinsed with distilled water and dripped before being washed three times with the water to be analyzed.

The sampling was carried out manually from the two fixed intake points. The bottle was held with one hand, immersed to a 20 cm depth, filled with water and hermetically closed. Then the samples were carefully and promptly transported to the laboratory after collection at a temperature of $4^\circ$C in order to ensure the physico-chemical parameters.

2.4 Chemical and physical analysis

The knowledge of some basic physical and chemical parameters offers a preliminary estimation of the water quality and the extent of eventual contamination. The physico-chemical parameters were determined by standard chemical and instrumental methods.

The suspended Matter (SM) in water after and before STEP rejections was determined using centrifugation. A sample of 100 ml of water is poured into the cup and centrifuged for 20 min at 3000 tr/min. The reaping cullet centrifuge is placed in a porcelain dish and dried to constant weight at 1050 C. After drying, the dish is placed in the desiccator for one hour and then it is weighed quickly.

Chemical Oxygen Demand (COD) is measured by colorimetric method after digestion of the sample for two hours in a COD reactor (DCO Spectrophotometer DREU2010 Adapteur de tube DCO sur DREL12010). When the biochemical oxygen demand (BD05) is defined as the amount of oxygen consumed speaks aerobic micro-organisms for the decomposition of biodegradable organic matter in the water to be analyzed for a period of five days.

The sample is stirred in a flask sealed incubator connected to a mercury manometer During biodegradation of organic matter, microorganisms consume oxygen from the air in the bottle causing a reduction in pressure above the sample. This vacuum is transmitted to the mercury manometer and oxygen consumption and read on the scale gauge.
So, we used colorimetric methods in order to quantify the major elements ($\text{NH}_4^+$, $\text{NO}_2^-$, $\text{NO}_3^-$). For each determination, a water volume of 20 ml was mixed in a cell with the suitable
reagent HACH. The unit was stirred and then left at rest for a pre-determined time to allow the complete colour-making reaction of the analyte with the reagent. The cell was then placed in a calibrated spectrophotometer, previously set at the wavelength of maximum absorbance. The quantification of the heavy metals lead (Pb) and cadmium (Cd) and selenium (Se) was carried out by a Lovibond spectrophotometer. The methods of analysis of the remaining parameters are gathered in Table 1.

<table>
<thead>
<tr>
<th>Temperature</th>
<th>Thermometer</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>Potentiometry</td>
</tr>
<tr>
<td>COD</td>
<td>Reflux method</td>
</tr>
<tr>
<td>BDO&lt;sub&gt;5&lt;/sub&gt;</td>
<td>5- days incubation</td>
</tr>
<tr>
<td>Suspended matter</td>
<td>Gravimetric</td>
</tr>
<tr>
<td>Nitrates</td>
<td>Colorimetric method</td>
</tr>
<tr>
<td>Nitrites</td>
<td>Colorimetric method</td>
</tr>
<tr>
<td>Ammonium ions</td>
<td>Colorimetric method</td>
</tr>
<tr>
<td>Lead</td>
<td>Photometric method</td>
</tr>
<tr>
<td>Cadmium</td>
<td>Photometric method</td>
</tr>
<tr>
<td>Selenium</td>
<td>Photometric method</td>
</tr>
</tbody>
</table>

Table 1. Water quality test methods.

### 3. Results and discussion

The results of the physico-chemical analyses of the Medjerda water before and after the STEP effluent input (site E and A respectively) are reported on Tables 2 and 3 respectively.

<table>
<thead>
<tr>
<th>Period</th>
<th>T (°C)</th>
<th>pH</th>
<th>SM (mg/l)</th>
<th>COD (mg/l)</th>
<th>BDO&lt;sub&gt;5&lt;/sub&gt; (mg/l)</th>
<th>NO&lt;sub&gt;3&lt;/sub&gt; (mg/l)</th>
<th>NO&lt;sub&gt;2&lt;/sub&gt; (mg/l)</th>
<th>NH&lt;sub&gt;4&lt;/sub&gt; (mg/l)</th>
<th>Pb (mg/l)</th>
<th>Cd (mg/l)</th>
<th>Se (mg/l)</th>
</tr>
</thead>
<tbody>
<tr>
<td>March</td>
<td>18.5</td>
<td>7.60</td>
<td>19</td>
<td>11</td>
<td>1.4</td>
<td>0.006</td>
<td>0.002</td>
<td>0.26</td>
<td>0.008</td>
<td>0.01</td>
<td>0.08</td>
</tr>
<tr>
<td>April</td>
<td>18.5</td>
<td>7.39</td>
<td>25.5</td>
<td>82.2</td>
<td>4.6</td>
<td>5.4</td>
<td>0.04</td>
<td>0.11</td>
<td>0.76</td>
<td>0.09</td>
<td></td>
</tr>
<tr>
<td>May</td>
<td>19.7</td>
<td>7.77</td>
<td>31.6</td>
<td>9</td>
<td>2.6</td>
<td>1.9</td>
<td>0.003</td>
<td>0.12</td>
<td>0.85</td>
<td>0.09</td>
<td></td>
</tr>
<tr>
<td>June</td>
<td>20.3</td>
<td>7.81</td>
<td>40.5</td>
<td>25</td>
<td>15</td>
<td>1.7</td>
<td>0.002</td>
<td>0.12</td>
<td>1.18</td>
<td>0.10</td>
<td></td>
</tr>
<tr>
<td>July</td>
<td>24.8</td>
<td>7.52</td>
<td>55</td>
<td>37</td>
<td>19</td>
<td>0.7</td>
<td>0.001</td>
<td>0.13</td>
<td>1.44</td>
<td>0.18</td>
<td></td>
</tr>
<tr>
<td>Limiting values</td>
<td>30</td>
<td>7.9</td>
<td>30</td>
<td>120</td>
<td>40</td>
<td>50</td>
<td>0.1</td>
<td>0.5</td>
<td>0.03</td>
<td>0.01</td>
<td></td>
</tr>
</tbody>
</table>

Table 2. Water parameters at site E (upstream STEP).

<table>
<thead>
<tr>
<th>Period</th>
<th>T (°C)</th>
<th>pH</th>
<th>SM (mg/l)</th>
<th>COD (mg/l)</th>
<th>BDO&lt;sub&gt;5&lt;/sub&gt; (mg/l)</th>
<th>NO&lt;sub&gt;3&lt;/sub&gt; (mg/l)</th>
<th>NO&lt;sub&gt;2&lt;/sub&gt; (mg/l)</th>
<th>NH&lt;sub&gt;4&lt;/sub&gt; (mg/l)</th>
<th>Pb (mg/l)</th>
<th>Cd (mg/l)</th>
<th>Se (mg/l)</th>
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</thead>
<tbody>
<tr>
<td>March</td>
<td>18.2</td>
<td>7.47</td>
<td>66.5</td>
<td>118</td>
<td>45.8</td>
<td>0.7</td>
<td>0.26</td>
<td>2.64</td>
<td>0.04</td>
<td>0.123</td>
<td></td>
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<tr>
<td>April</td>
<td>18.4</td>
<td>7.19</td>
<td>90.5</td>
<td>296</td>
<td>47</td>
<td>9.4</td>
<td>0.68</td>
<td>5.92</td>
<td>1.84</td>
<td>0.133</td>
<td></td>
</tr>
<tr>
<td>May</td>
<td>20.3</td>
<td>7.50</td>
<td>109.6</td>
<td>132</td>
<td>33</td>
<td>2.6</td>
<td>0.37</td>
<td>6.82</td>
<td>1.86</td>
<td>0.169</td>
<td></td>
</tr>
<tr>
<td>June</td>
<td>21</td>
<td>7.88</td>
<td>154.5</td>
<td>144</td>
<td>75</td>
<td>2.3</td>
<td>0.14</td>
<td>7.14</td>
<td>1.95</td>
<td>0.172</td>
<td></td>
</tr>
<tr>
<td>July</td>
<td>25.1</td>
<td>7.72</td>
<td>155</td>
<td>158</td>
<td>145</td>
<td>1.9</td>
<td>0.006</td>
<td>11.92</td>
<td>2.08</td>
<td>0.280</td>
<td></td>
</tr>
</tbody>
</table>

Table 3. Water parameters at site A (downstream STEP).
3.1 Temperature

The water temperature plays an important role in the solubility of salts and gases. As reported in (figure 3) the maximal temperatures were recorded in July, specifically 25.1 °C at site A and 24.8 °C at site E. The temperatures observed at the both was lower than the Algerian standards; we may recall that temperatures higher than 15°C favor the development of micro-organisms and in the same time intensify the organo-leptical parameters like odor and activate chemical reactions (Athmani, A. S. 2005), (Guasmi, 2004), (Botton et al. 1999), (Meinck et al. 1977).

![Temperature variations before and after the intake of STEP effluent as a function of time.](image)

3.2 pH

The pH is also a factor which influences the biological activity of the water micro flora. As seen on (figure 4) all samples from both sites display a slightly basic pH, ranging from 7.4 to 7.9 in the first site and from 7.1 to 7.9 in the second one. Generally, the obtained values were in agreement with the Algerian and the word health organization standards of 7.0 to 8.5. A neutral pH in the interval (6.5 - 8.5) characterizes water where life develops in optimal way (Larpent, 1997), (Guasmi & Djabri, 2006).

3.3 Suspended matters

To appreciate a water quality; it is always necessary to estimate quantitatively its load in dissolved and particular matter. As shown in (Figure. 5) we notice that the contents in
suspended matter (SM) are more raised in the site A than in the site E and the values vary from 19 to 55 (mg/g) in the first one and from 66.5 to 155 in the second.

![pH variations chart]

Fig. 4. pH variations before and after the intake of STEP effluent as a function of time.

We also noticed that the suspended matter content varies significantly with the season and the rates are higher in dry period than in pluvial one.

As soon as suspended matter content exceeds the Algerian standards in site A and that can induce the accumulation of higher quantities of toxic substances such as metals, pesticides, mineral oils, polycyclic aromatic hydrocarbons (Tiri et al. 2007), (Derradji et al. 2007), (Abulude et al. 2007), (Bouchenafa et al. 2008) which may have an adverse effect on water quality.

3.4 Chemical oxygen demand (COD) and biological demand of oxygen (BDO5)

The Chemical Oxygen Demand (COD) is the amount of oxygen consumed in mg/l, for oxidable materials in an effluent. It is representative of most of organic compounds but also oxidable inorganic salts (sulfides, chlorides...).

As shown in (Figures. 6) we noticed that values of chemical oxygen demand are lower than the Algerian standards in the site E but higher in the site A. We also note that most of the values exceed the Algerian standards in the site A for a maximum of 296 mg/l. This allows to classifying the water of this site as very bad quality. On the other hand, the high temperatures of the warmest months and also the dry periods of the year, seem to increase significantly the COD of these water.
Fig. 5. Total suspended solids variations before and after the intake of STEP effluent as a function of time.

Fig. 6. Variations of chemical oxygen demand before and after the intake of STEP effluent as a function of time.
The biochemical oxygen demand (BDO$_5$) allows the evaluation of the present biodegradable organic matters in water. (Figure 7) showed that the BDO$_5$ values are lower than the Algerian standards in the site E but exceed them in the site E only in dry period and this is in agreement with the results obtained by Guasmi et Djabri...etc (Guasmi & Djabri, 2006), (Tiri et all. 2007), (Benza et all. 2005), (Guasmi & Djabri, 2005).

Fig. 7. Variations of biochemical oxygen demand before and after the intake of STEP effluent as a function of time.

### 3.5 Dissolved nitrogen

Human activities and mainly those related to agriculture are a major cause of the presence of nitrates and nitrites in surface water. These two substances are responsible for many problems not only for environment but also for human health. Indeed, although not directly toxic, they participate in eutrophication phenomena of surface water.

The made analyses allow to notice that the rates of nitrates obtained in the two sites are lower than the standards required which are in the order of 50 mg/l (Figures 8, 9 and 10). On the other hand the results concerning nitrites and ammonium ion show superior rates to the standards required (0,1 mg / l) at the site A. Their presence can be explained by an incomplete oxidation of the ammonia water; or a nitrate reduction reaction. This pollution can be caused by intense agricultural activity (the region of study is known for its agricultural vocation) and misuse of chemical fertilizers around the sewage waste water and Oued Medjerda.
Fig. 8. Nitrates variations before and after the intake of STEP effluent as a function of time.

Fig. 9. Nitrites variations before and after the intake of STEP effluent as a function of time.
The highest contents of nitrates and nitrites are observed in April and may be explained by the leaching by rainfall observed at this precipitation period. The highest ammonium concentration is observed in July and may be explained by a dilution effect, the water volume flowing in the river being much lower in this dry period (Derradji et all. 2007), (Abulude et all. 2007), (Bouchenafa et all. 2008).

### 3.6 Metals

It has been clearly demonstrated that serious environmental problems were caused when the waste or the treated water may contain high levels of trace elements. These elements are potentially toxic to plants and human beings. However the case of selenium is complex since it is at the same time a trace element; essential to life; and a toxic; and this in a narrow range of concentration.

According to the results obtained (figure 11 and 12) we noticed that lead (Pb) levels Varied between 0,01 to 1,44 mg/l in the first site and from 0,04 to2,08 mg/l in the second one which are within acceptable limit value of 0,03 mg/l. Similarly for cadmium (Cd), its concentration varies from 0,08 to 0,18 mg/l in the site E and from 0,11 to 0,3 mg/l in the site A, whereas the limit value is 0,01 mg/l. while this is not the case for selenium (Se). For which we noted its non-existence.

So we can say that there is a generated pollution by metals such as lead and cadmium. With high levels in the site A and less in the site E. The high content in of metals is probably due to the direct pollution of the rivers by the industrial wastes; atmospheric precipitation involving a certain part of the pollutants present in the atmosphere and in consequence of the streaming of water on the grounds involving accumulated lead or cadmium at surface (Abulude et all. 2007), (Igbinosa & Okoh, 2009).
Fig. 11. Lead variations before and after the intake of STEP effluent as a function of time.

Fig. 12. Cadmium variations before and after the intake of STEP effluent as a function of time.
4. Conclusion

Medjerda is the principal river in the north-east of Algeria. This study examines the impacts of discharges of sewage wastewater (STEP Henencha) on the quality of this river. Monitoring was made upstream and downstream the STEP rejections from March to July 2009. Parameters measured include pH, temperature, total suspended matters, chemical oxygen demand, biological oxygen demand, nitrate, nitrite, ammonium ions and trace elements (cadmium, lead and selenium) using standard methods. Unacceptably, high levels of the assayed parameters were observed downstream the STEP in many cases for chemical oxygen demand (118-296 mg/l), biological oxygen demand (33-145 mg/l), nitrite (0.14-0.68 mg/l), ammonium ions (2.64-11.92 mg/l); lead (0.04-2.08 mg/l) and cadmium (0.1-0.2 mg/l) during the study period and outside the compliance levels of Algerian guidelines. We can also conclude that the lowest levels of metal ions were registered during flood periods and this is certainly due to the high dilution facing the river following rainfall. The results obtained confirmed that there was an adverse impact on the physico-chemical characteristics and this implies that the water of this river has a major pollution problem.

5. Abbreviations

STEP Henancha : Wastewater treatment station of Henancha
COD: Chemical oxygen demand
BDO₅: Biological oxygen demand
SM : Suspended matter
NH₄⁺: Ammonium ions
NO₂⁻: Nitrites
NO₃⁻: Nitrates
Pb: Lead.
Cd: Cadmium.
Se: Selenium.

6. References

Athmani, A. S. (2005). Evaluation de la qualité des eaux de surface, cas du bassin versant Oued Medjerda (Souk Ahras, Algérie), Thèse de Magister, Centre universitaire de Souk Ahras


Guasmi, I. (2004). Dégradation de la qualité de l’eau dans le bassin versant de l’Oued Medjerda (Souk Ahras, Algérie), Thèse de Magister, Université de Annaba, Algérie


The book attempts to cover the main fields of water quality issues presenting case studies in various countries concerning the physicochemical characteristics of surface and groundwaters and possible pollution sources as well as methods and tools for the evaluation of water quality status. This book is divided into two sections: Statistical Analysis of Water Quality Data; Water Quality Monitoring Studies.

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